

# Molecular Engineering of Solar-Powered Windows

Jacqueline Cole

Cavendish Laboratory, University of Cambridge, UK

ISIS Neutron and Muon Source, Rutherford Appleton Laboratory, UK

Argonne National Laboratory, USA



# Molecular Engineering of Solar-powered Windows



Jacqueline M. Cole

Cavendish Laboratory, University of Cambridge, UK

ISIS Neutron Facility, Rutherford Appleton Laboratory, UK

Argonne National Laboratory, USA



## Synopsis

- Scientific background and motivation
- Materials discovery using existing databases: pre-ADSP
- Auto-generating Custom Databases: ChemDataExtractor
- Materials discovery "Take #2" – The ADSP project
- Next Steps

# Background and Motivation

---

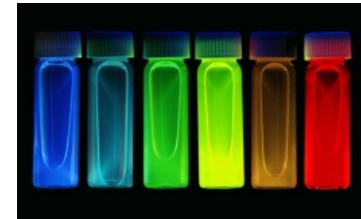
# Types of Solar Cells



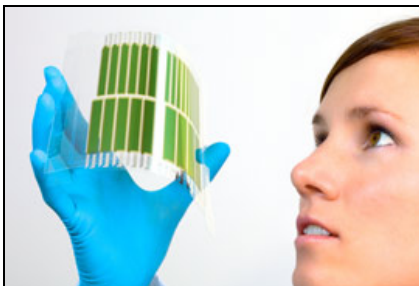
Silicon



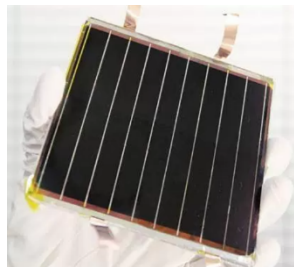
Thin-film Si



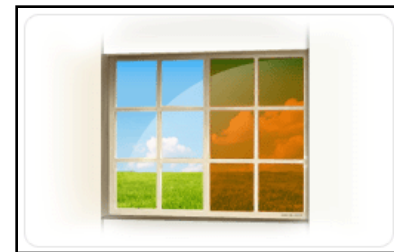
Quantum Dot



Organic PV



Perovskite PV



Dye-Sensitized  
Solar Cell (DSC)



# Solar-powered Windows



40% Total energy drain in the USA comes from buildings

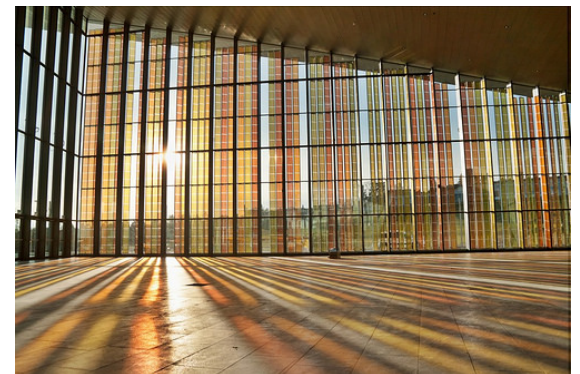
# DSCs in real-world action ...



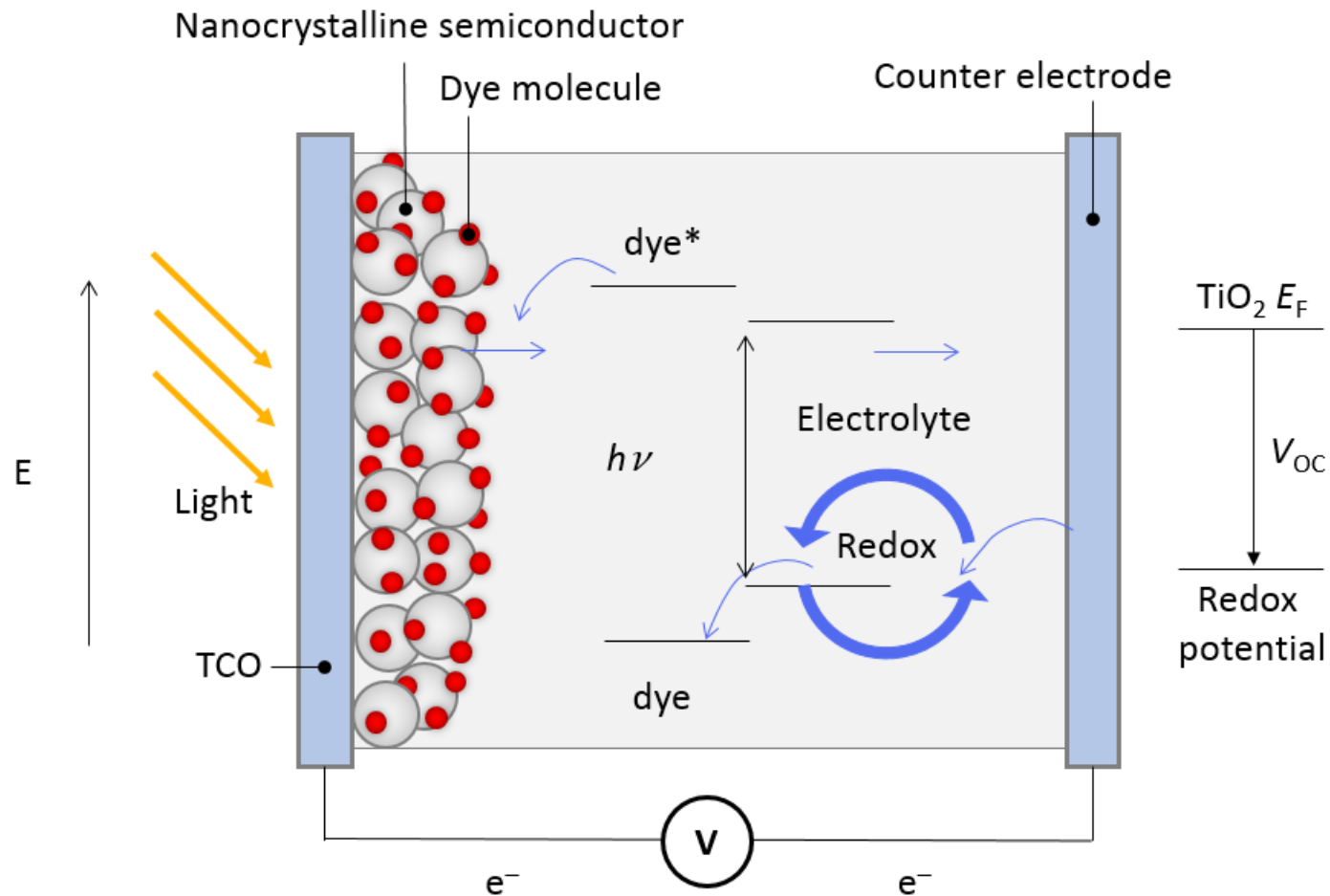
Graz Science Tower (Sep 17)



Swiss Tech  
Convention  
Centre



# Dye-sensitized solar cell



Economics: Price-to-Performance Ratio (and color options)

# Materials Discovery pre-ADSP

---



# Materials Discovery of DSC dyes

Volume 16 | Number 48 | 28 December 2014 | Pages 26393–27136

**PCCP**  
Physical Chemistry Chemical Physics  
www.rsc.org/pccp

ISSN 1463-9076

ROYAL SOCIETY OF CHEMISTRY

COMMUNICATION  
Jacqueline M. Cole et al.  
Data mining with molecular design rules identifies new class of dyes for dye-sensitized solar cells

Design

to

Device

$\text{TiO}_2$

redox

dye

$\text{e}^-$

**ADVANCED ENERGY MATERIALS**

M O L C U L A R

$\text{NR}_2$

$\text{COOH}$

$\text{NR}'_2$

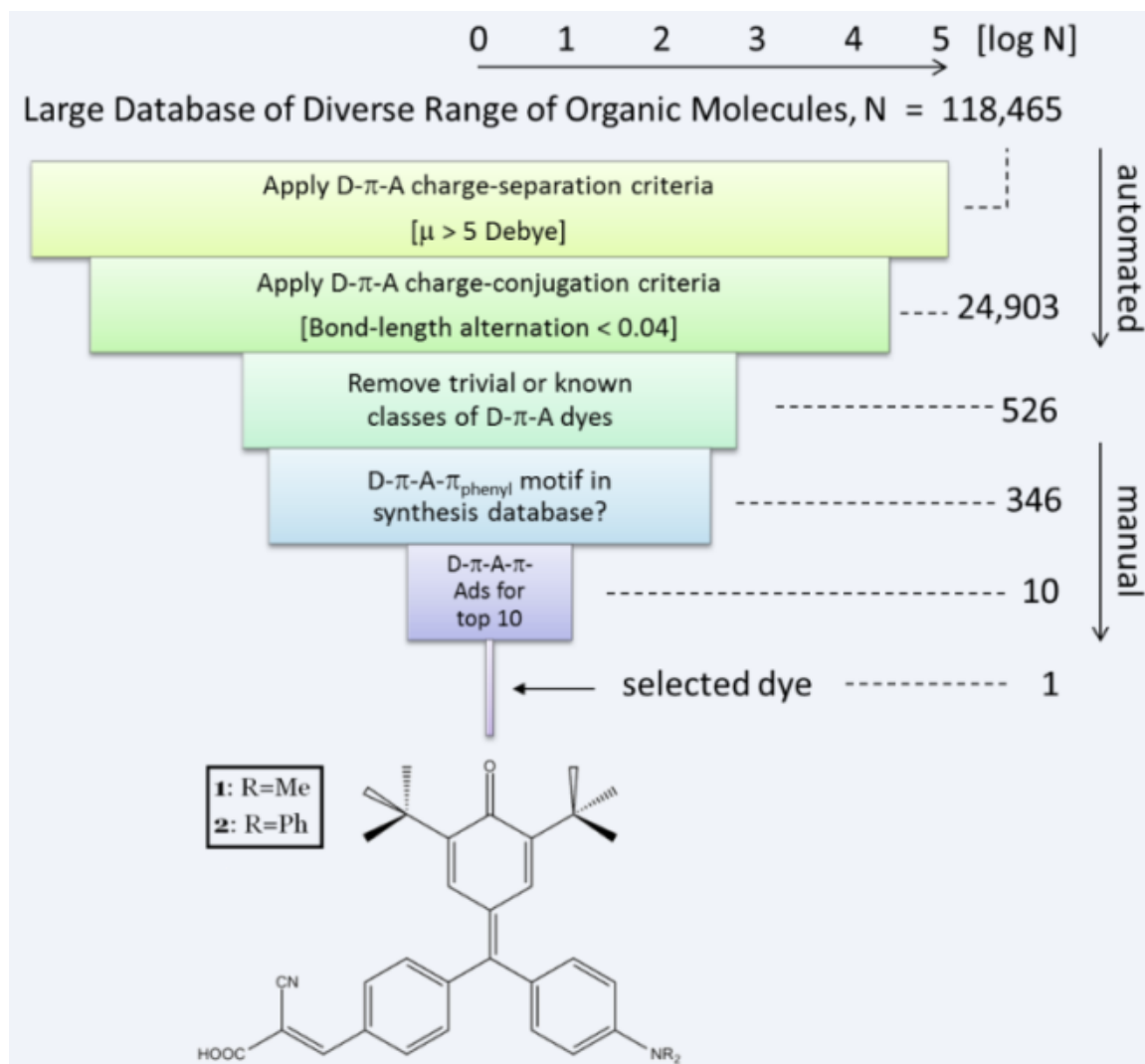
$\text{TiO}_2$

E N G I N E E R I N G

'Top-down' approach: Cole et al, PCCP, 16 (2014) 26684

'Bottom-up' approach: Cole et al, AEM 5 (2015) 1401728

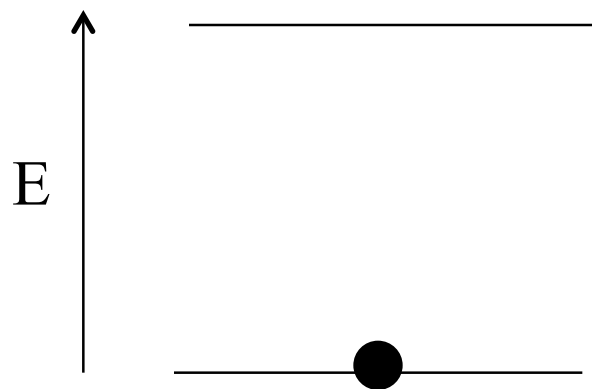
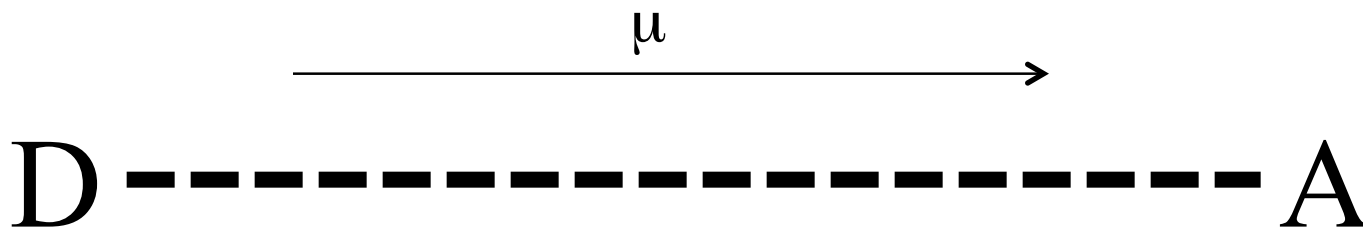
# Overarching Data-Mining Strategy



# Creating molecular design rules

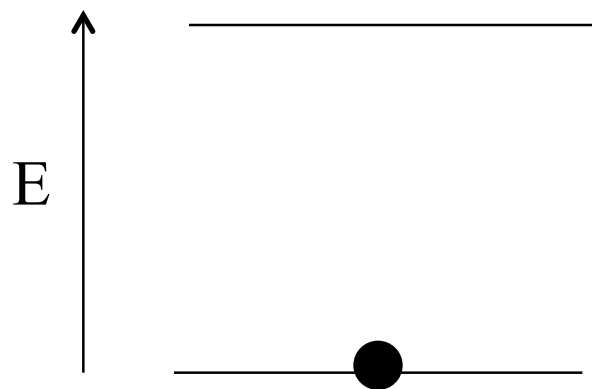
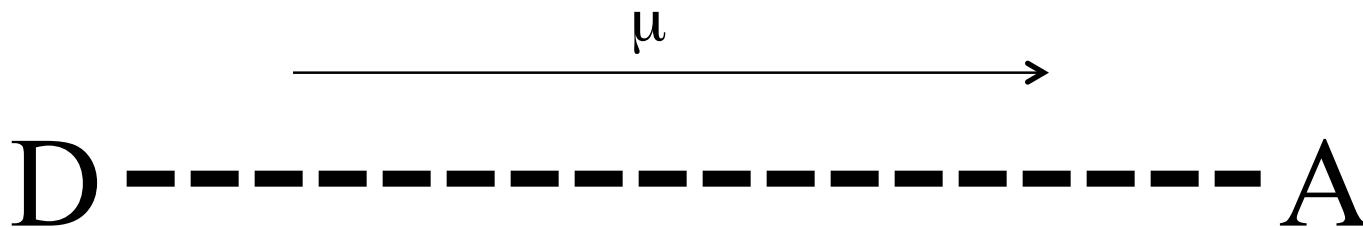
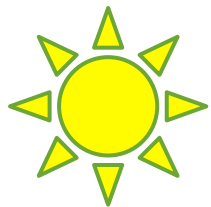
---

# Molecular Design Rules

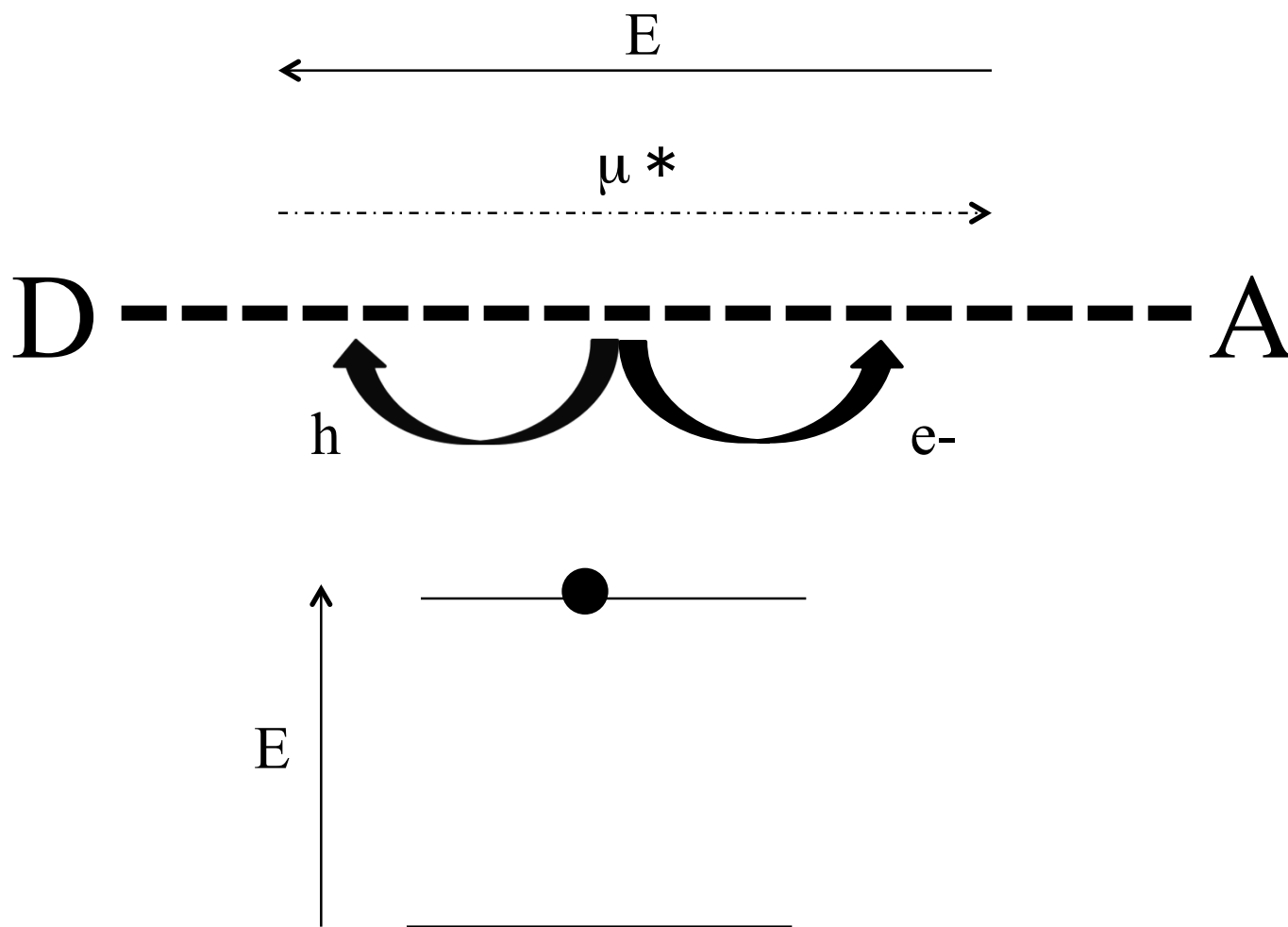
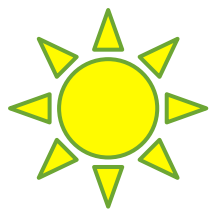




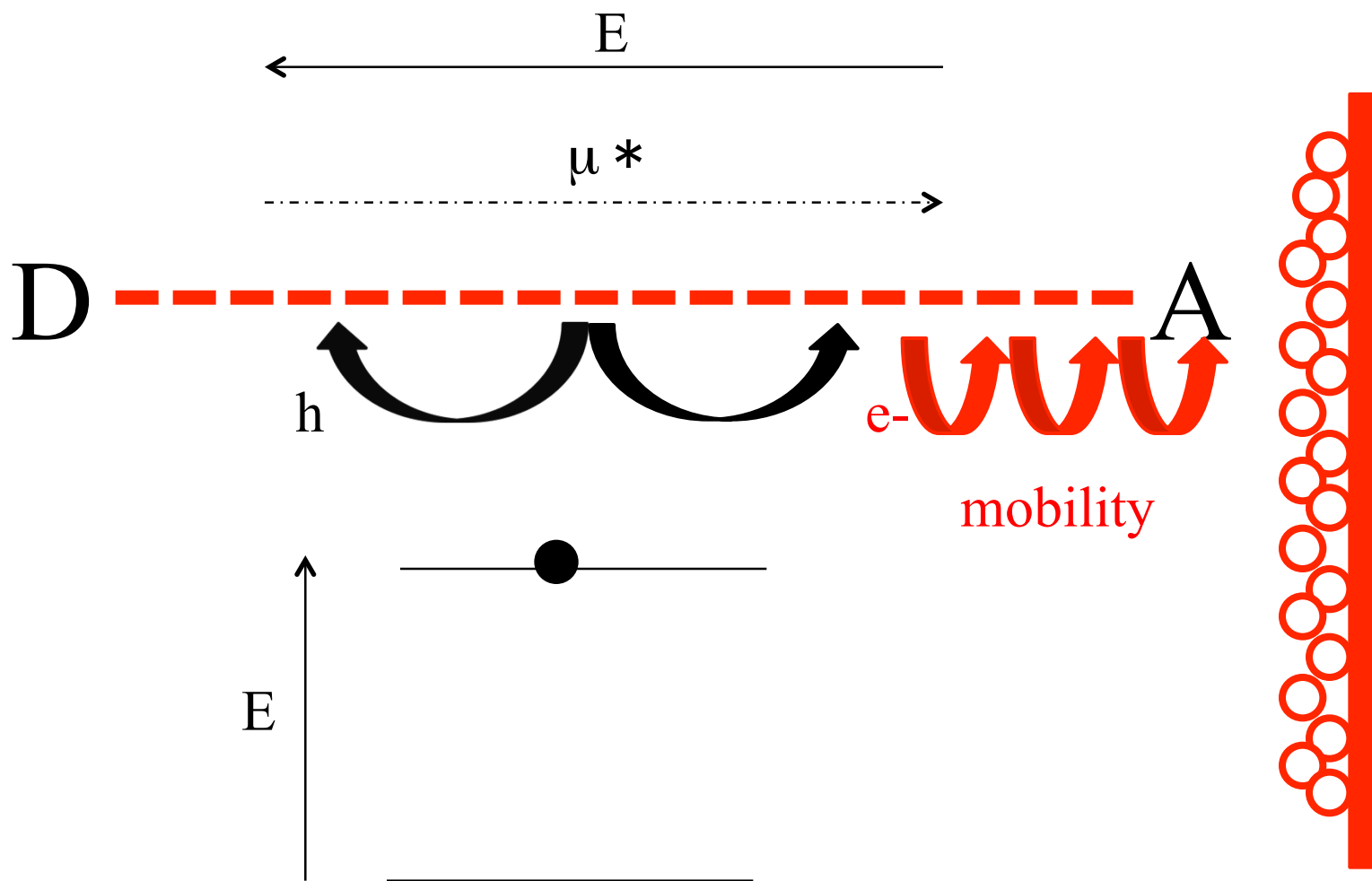
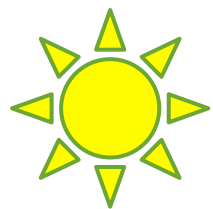
# Molecular Design Rules



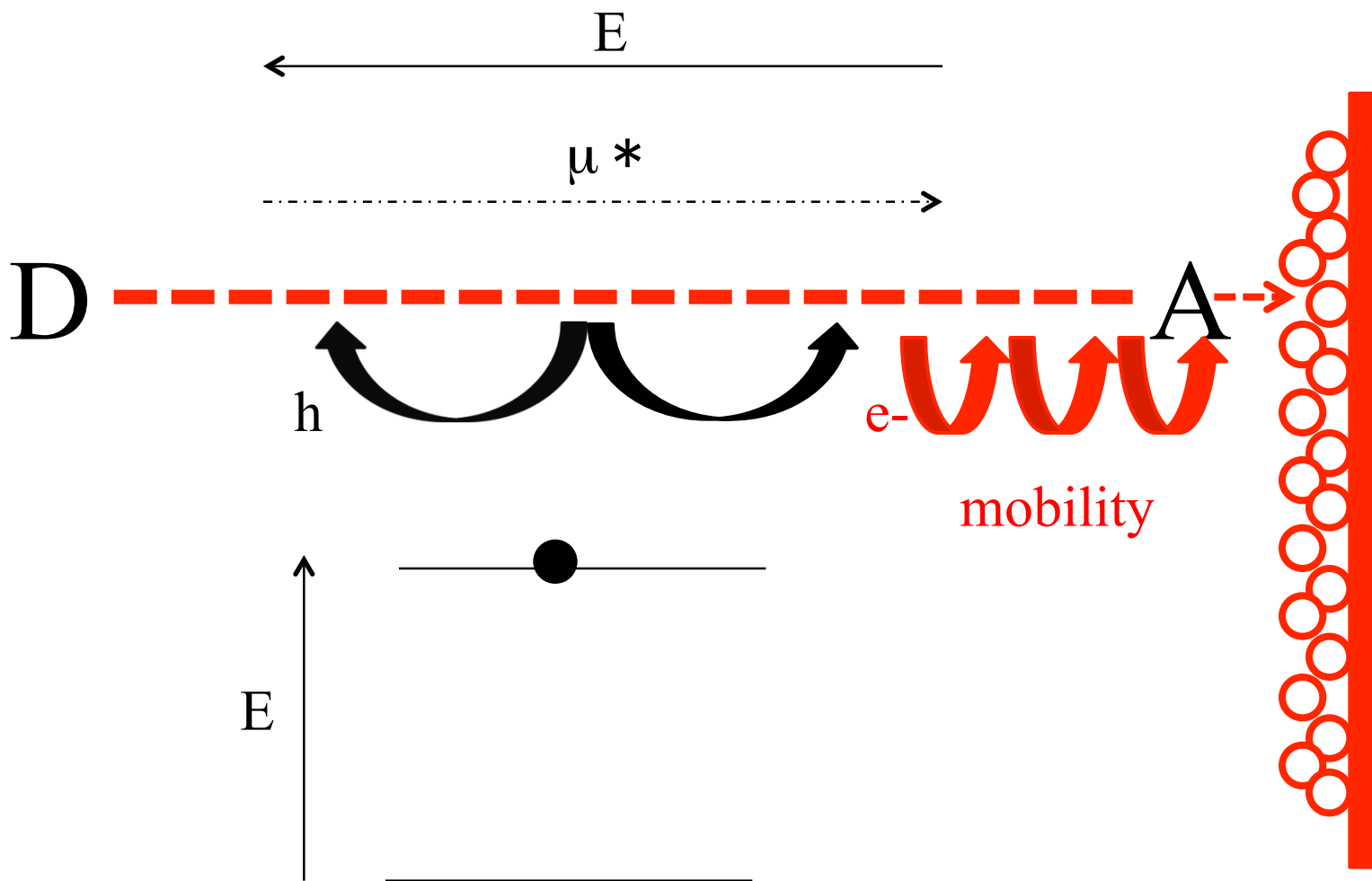
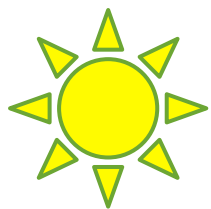
# Molecular Design Rules



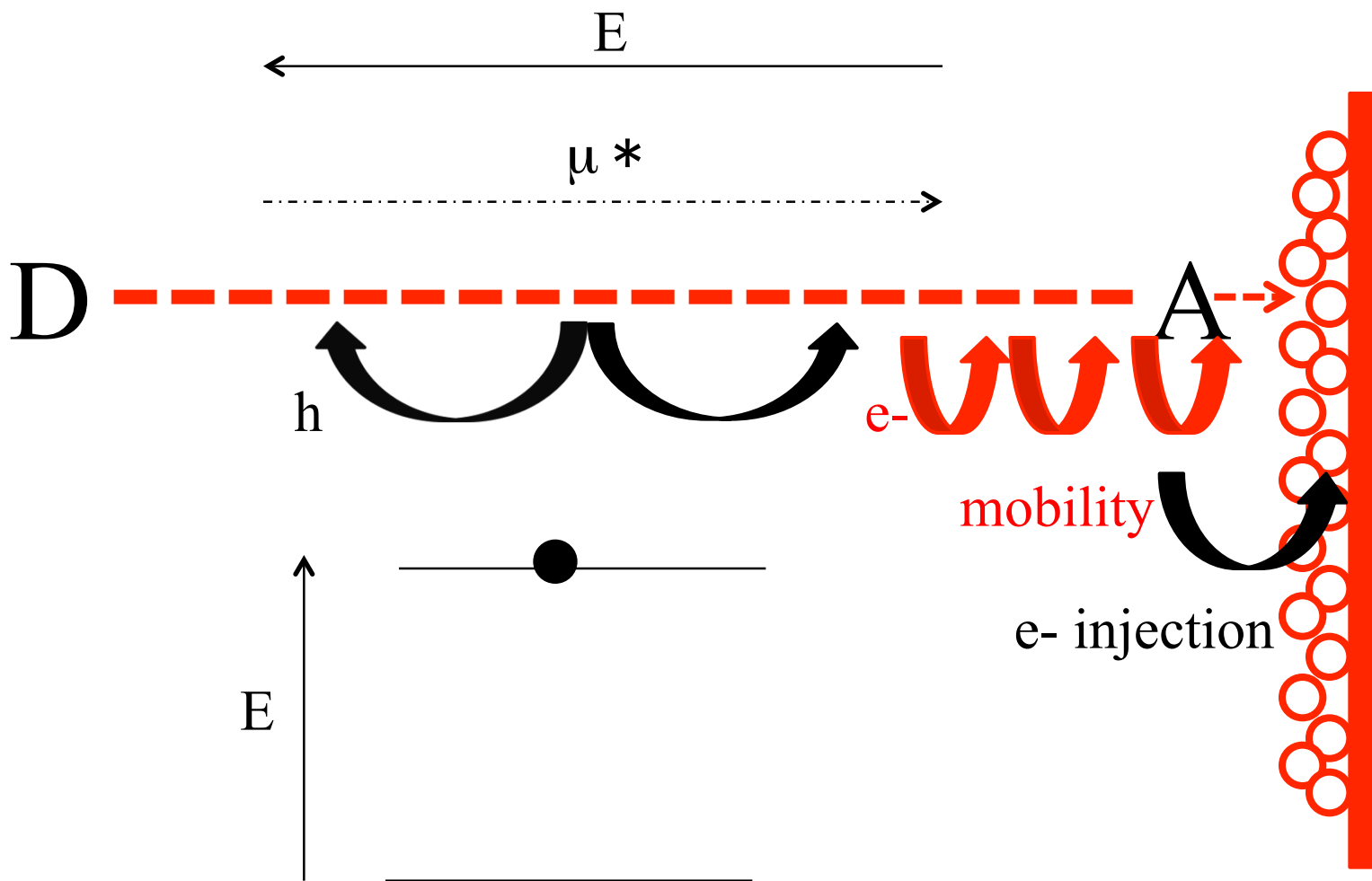
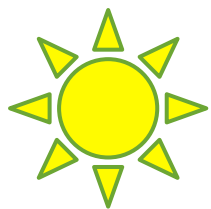
# Molecular Design Rules



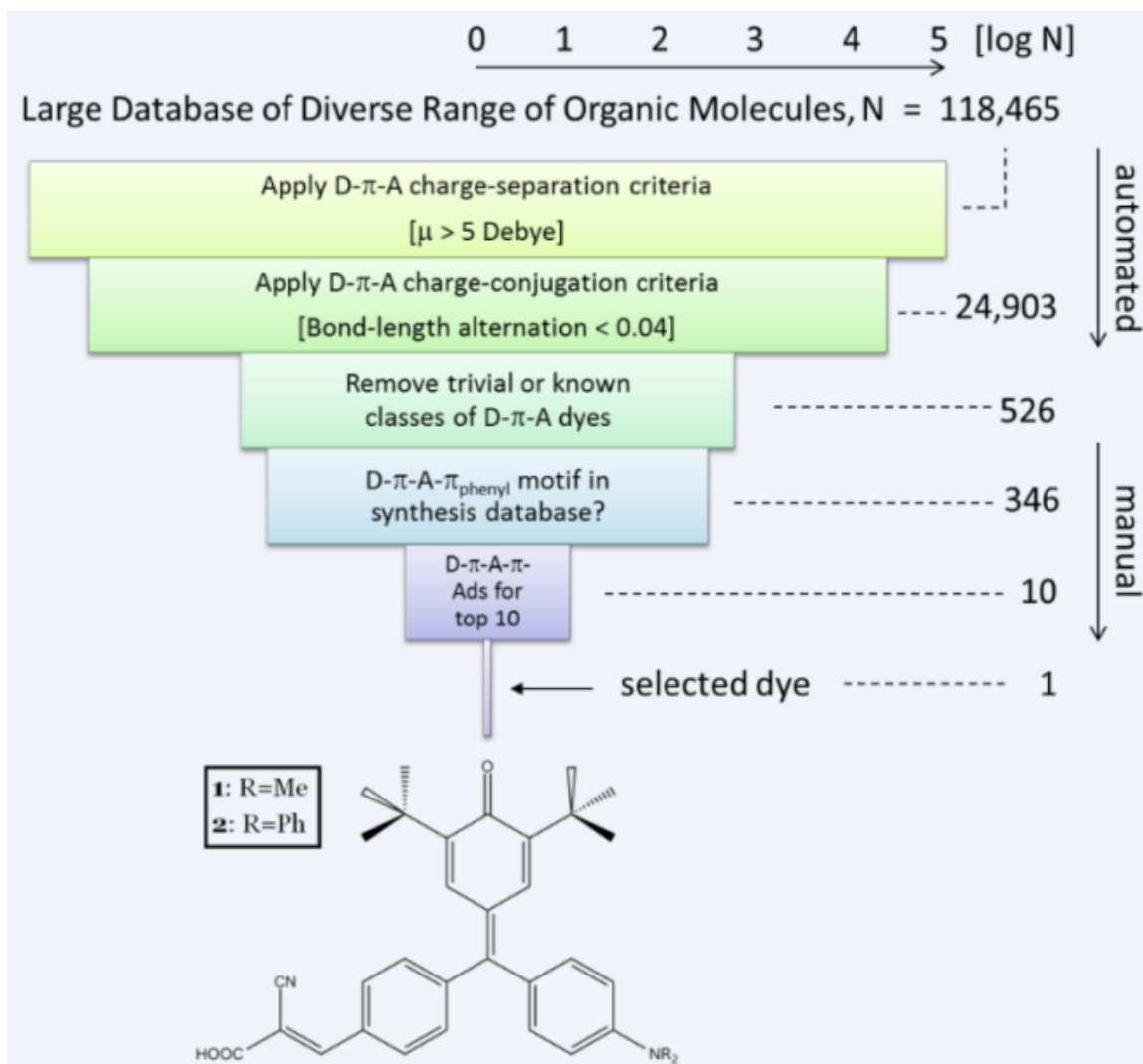
# Molecular Design Rules



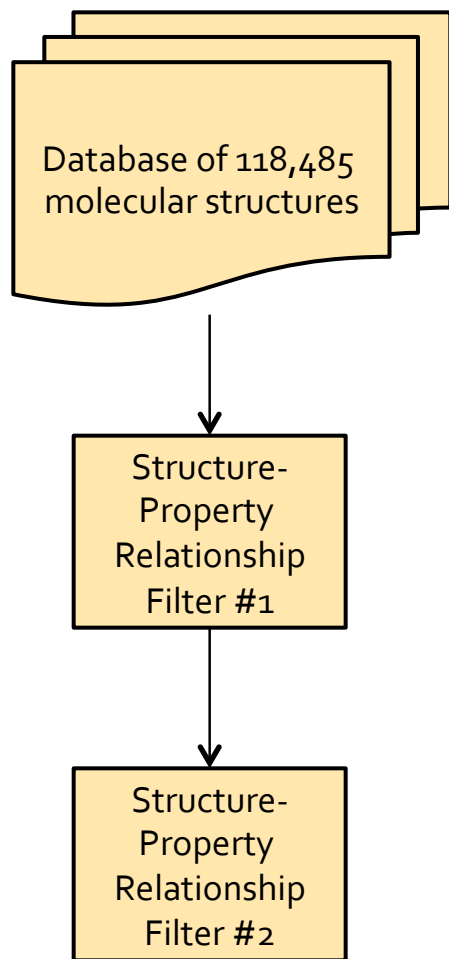
# Molecular Design Rules



# Overarching Data-Mining Strategy



# Auto-sorting a set of chemical space



A representative set of organic chemical space

#1 Apply D- $\pi$ -A charge-separation criterion

Calculate  $\mu$  for all 118,465 molecules

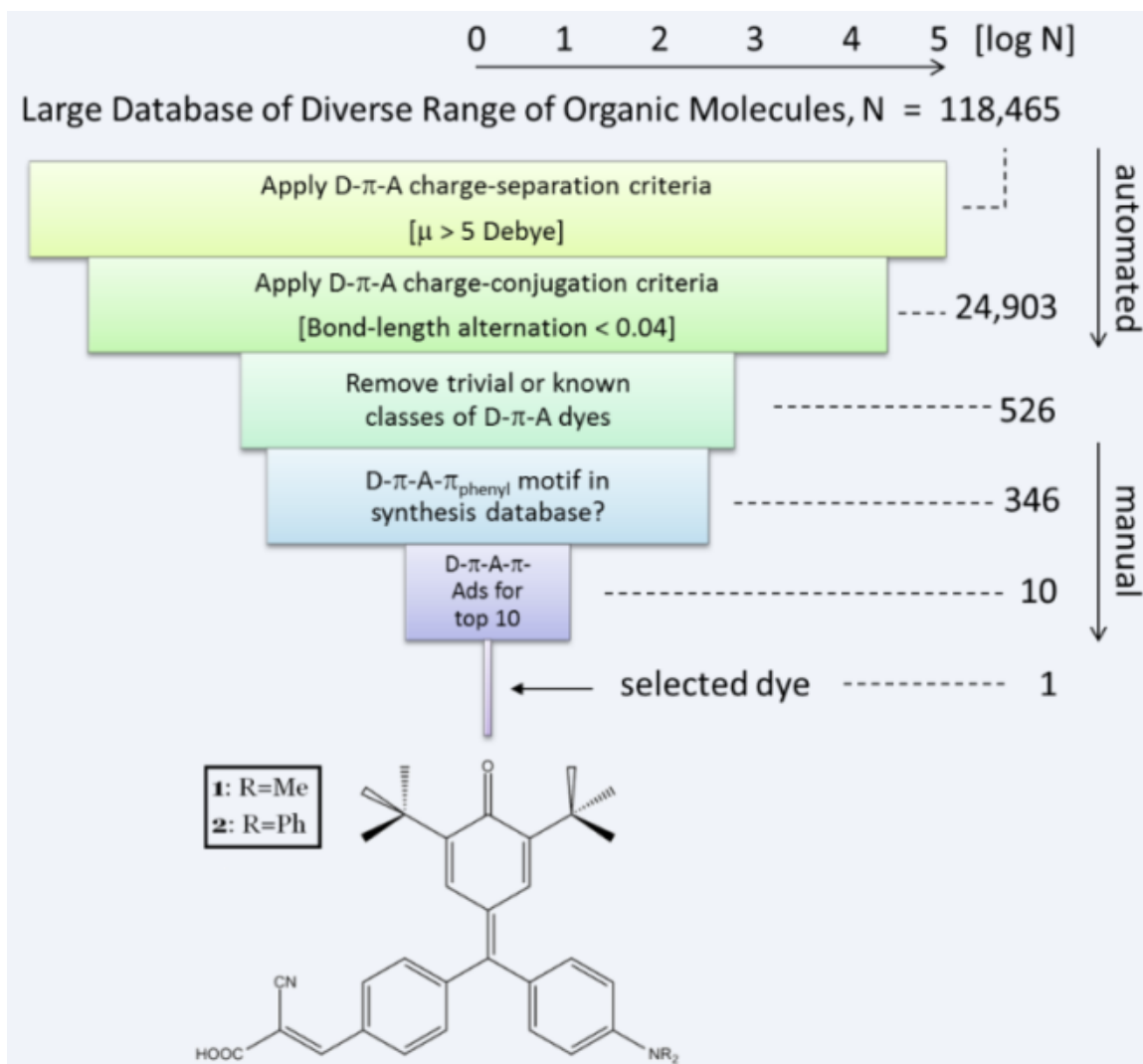
Set threshold:  $\mu > 5 D$

Semi-empirical PM7 in MOPAC

#2 Apply D- $\pi$ -A charge-conjugation criterion

Encode bond-length Alternation calculation

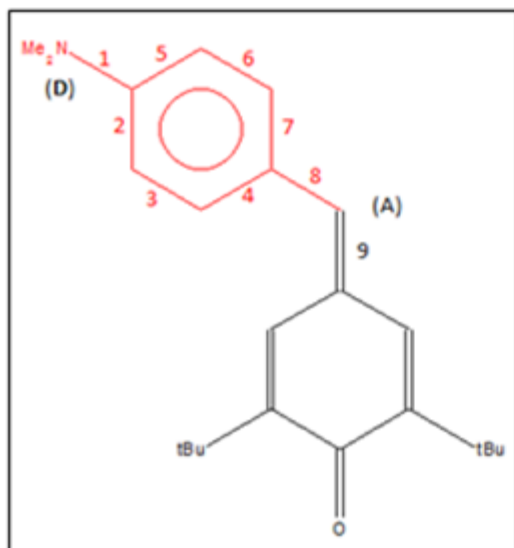
# Overarching Data-Mining Strategy



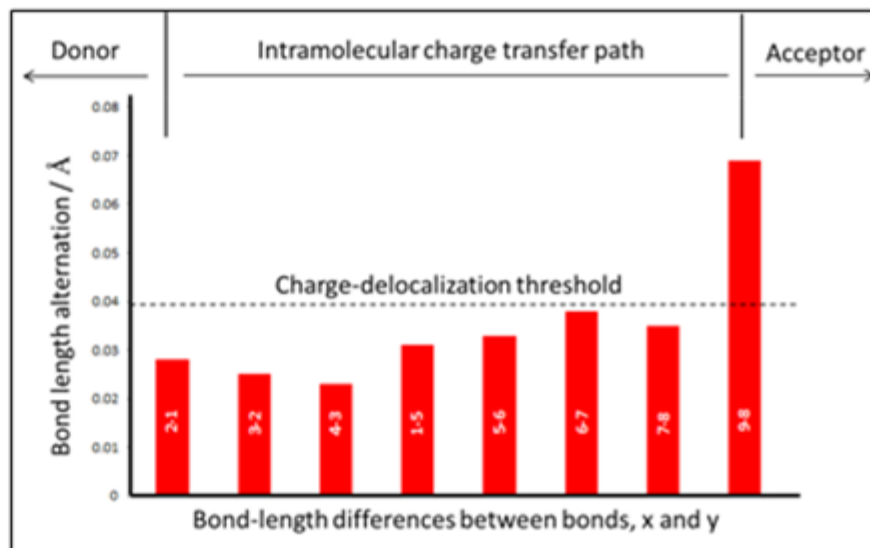


# Encoding D- $\pi$ -A Bond-length Alternation

Example Compound



BLA for each bond emanating from D (nitrogen)



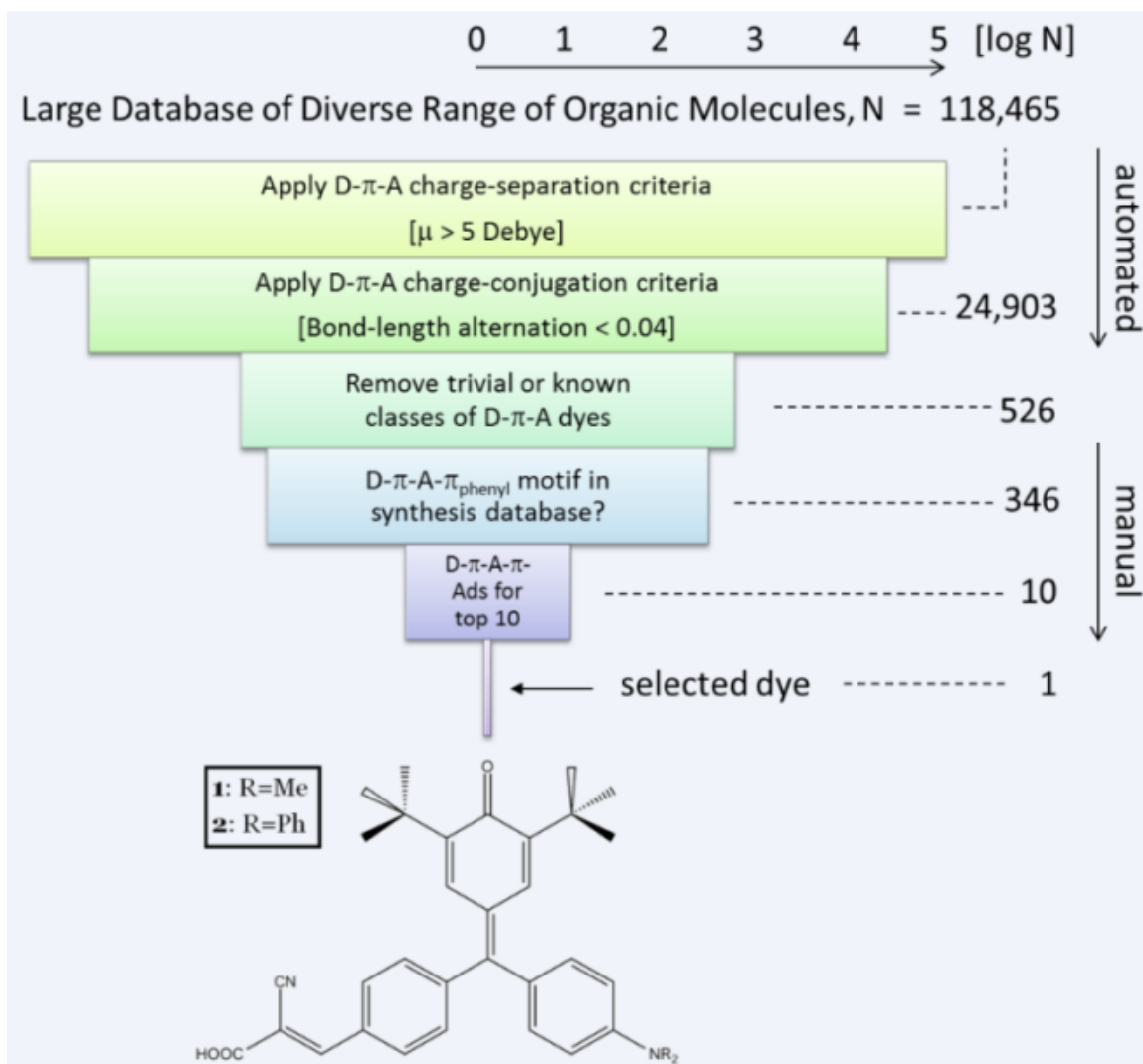
*Illustration of the BLA calculation of all bonds,  $n$ , in the red fragment of the example compound.*

The donor, D, is pre-selected from Hammett constants; here, D is the nitrogen on the N-Me<sub>2</sub> substituent.

The electron acceptor group begins at A, as identified by the individual BLA that first exceeds the BLA threshold, 0.04 Å, as bonds emanate from D. In this molecule, this BLA (0.069 Å) lies between bond 8 and 9.

The overall BLA for the D- $\pi$ -A molecular fragment is 0.0308 Å, *i.e.* significantly less than the threshold, BLA < 0.04 Å. Thus, this molecule passes the charge-delocalization criteria in the filtering algorithm.

# Overarching Data-Mining Strategy

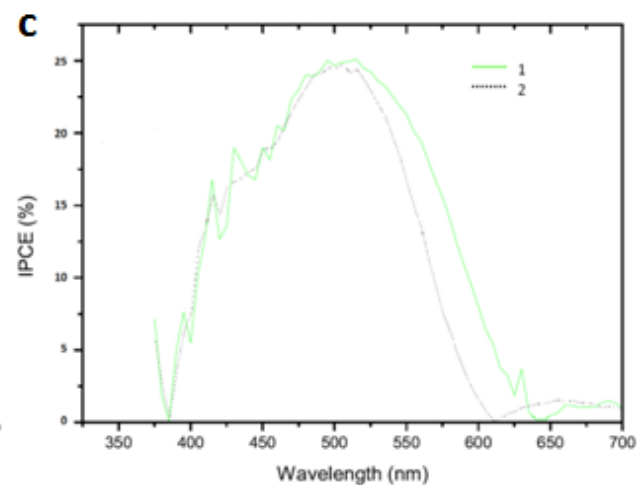
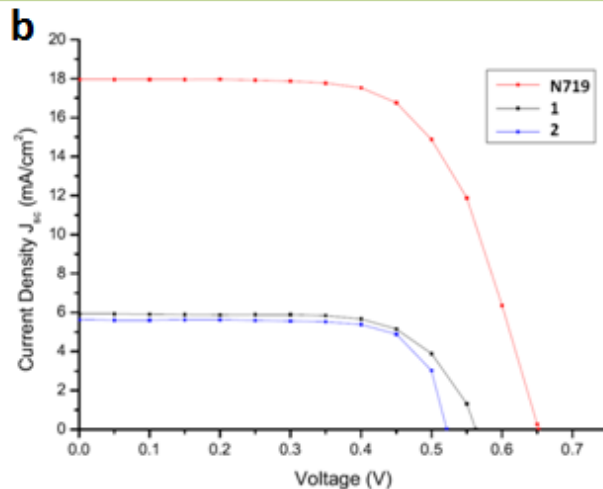
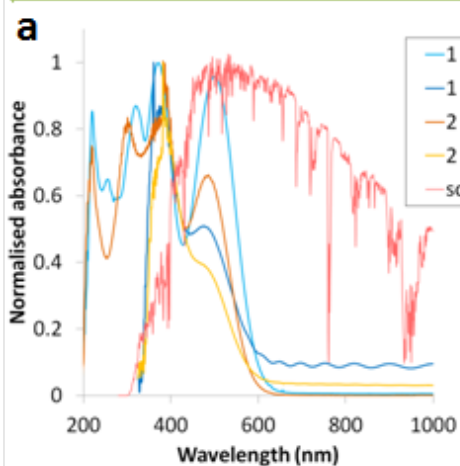


# Experimental Validation

---

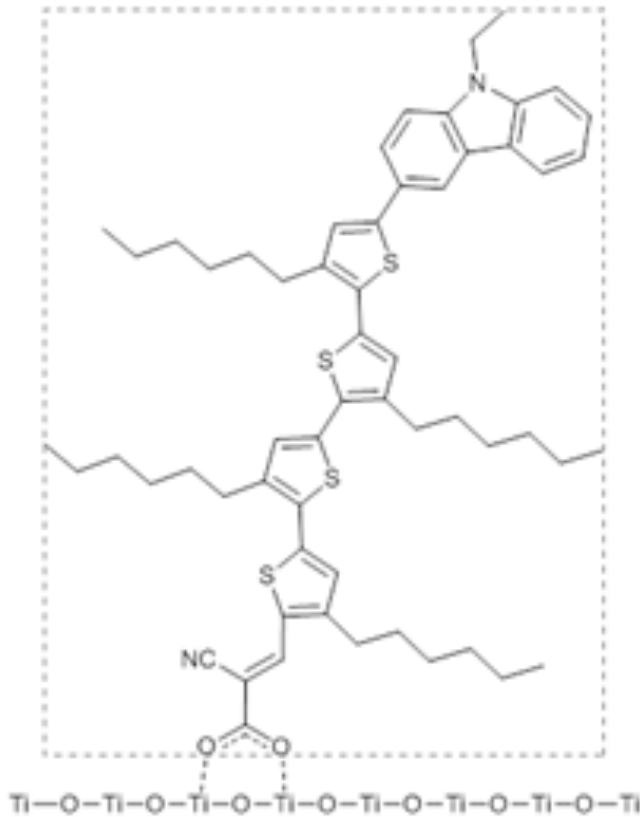
# DSC Device Testing

Dye	$V_{oc}$ (V)	$J_{sc}$ (mA/cm <sup>2</sup> )	FF	$\eta$ (%)	$\eta_{dye}:\eta_{N719}$ (%)
1	0.56	5.92	0.69	2.32	30.8
2	0.52	5.62	0.75	2.20	29.2
N719	0.65	17.96	0.64	7.54	100

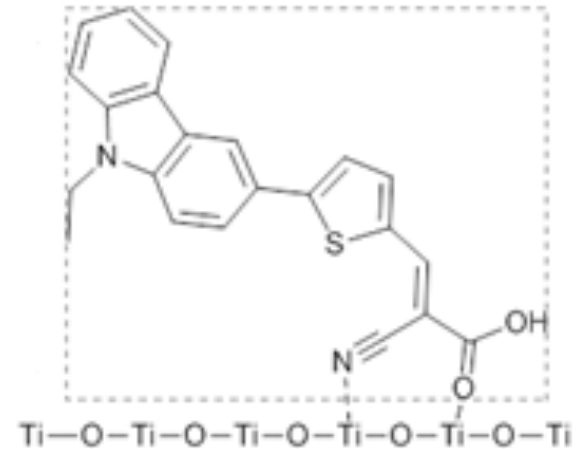


# Molecular Building Blocks of DSC Dyes

e.g.



MK-2 [PV eff: 7.7%]



MK-44 [PV eff: 1.8%]

# Materials discovery of DSC dyes

Molecular building blocks of predicted new classes of DSC dyes  
→ 1/3 world's best performance DSC dye in experimental validation

→ Far from a null result - Prediction strategy works!  
(and should improve with further refinement of data-mining method)

No significant dye aggregation effects observed in 1 or 2

1 and 2 exhibit unusually high melting points

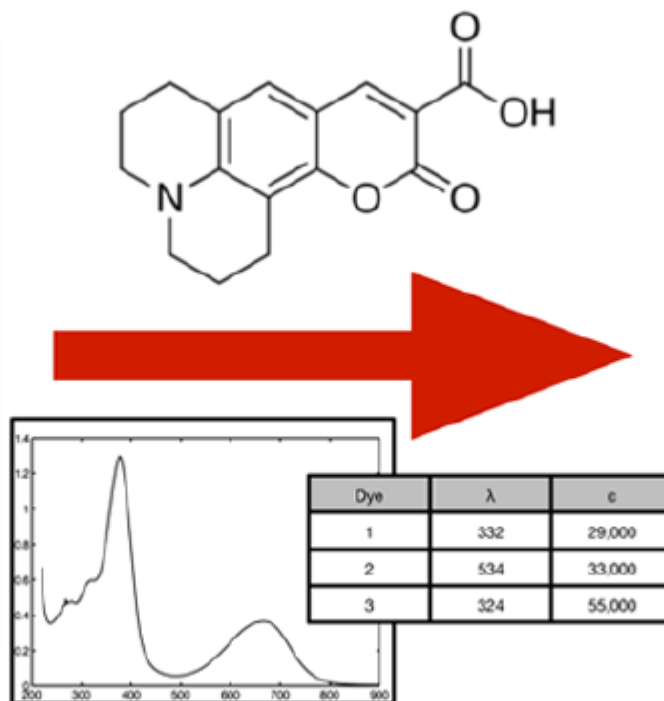
## What next?

Use custom-made material databases?

# Database Auto-generation tools

---

# ChemDataExtractor

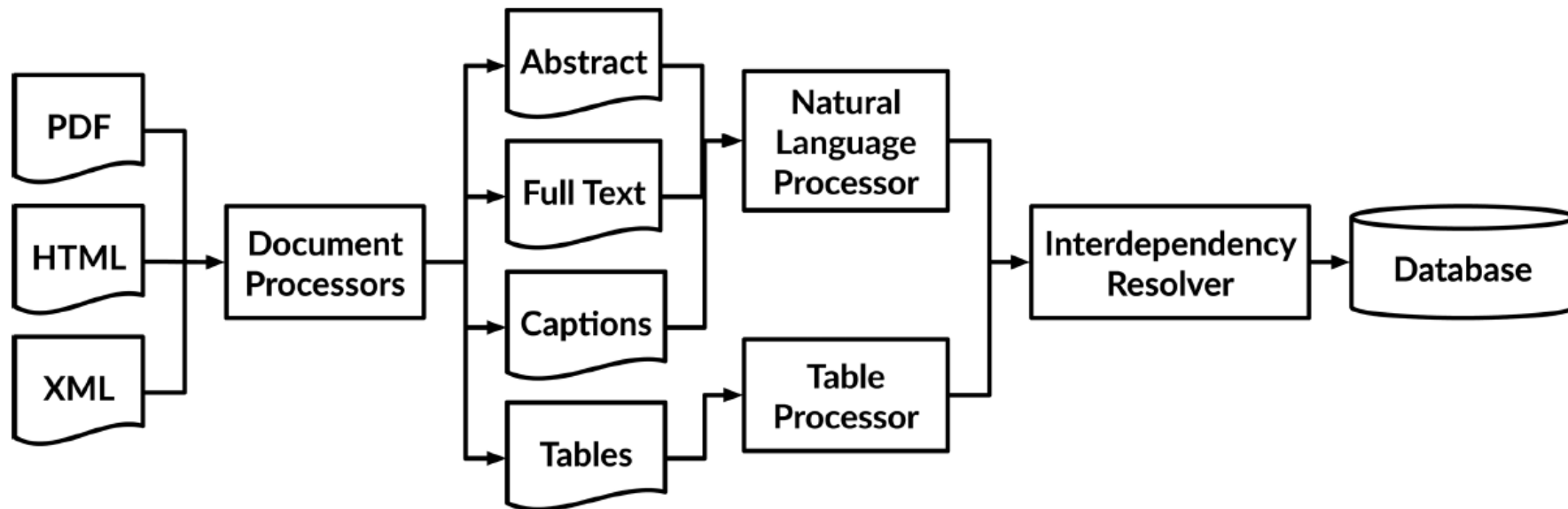


[www.chemdataextractor.org](http://www.chemdataextractor.org)

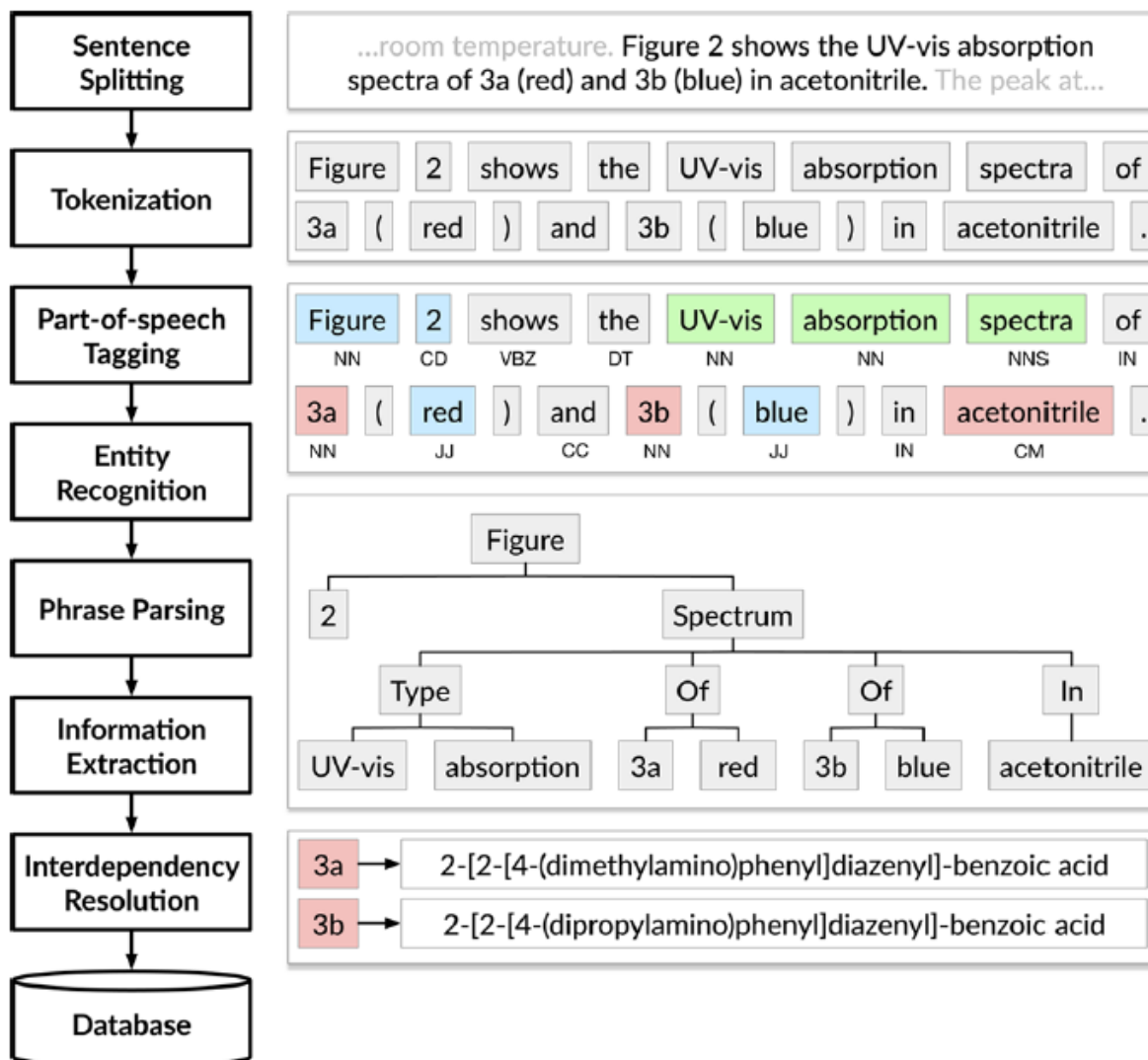
M. C. Swain, J. M. Cole *J. Chem. Inf. Model.* 56 (2016) 1894-1904



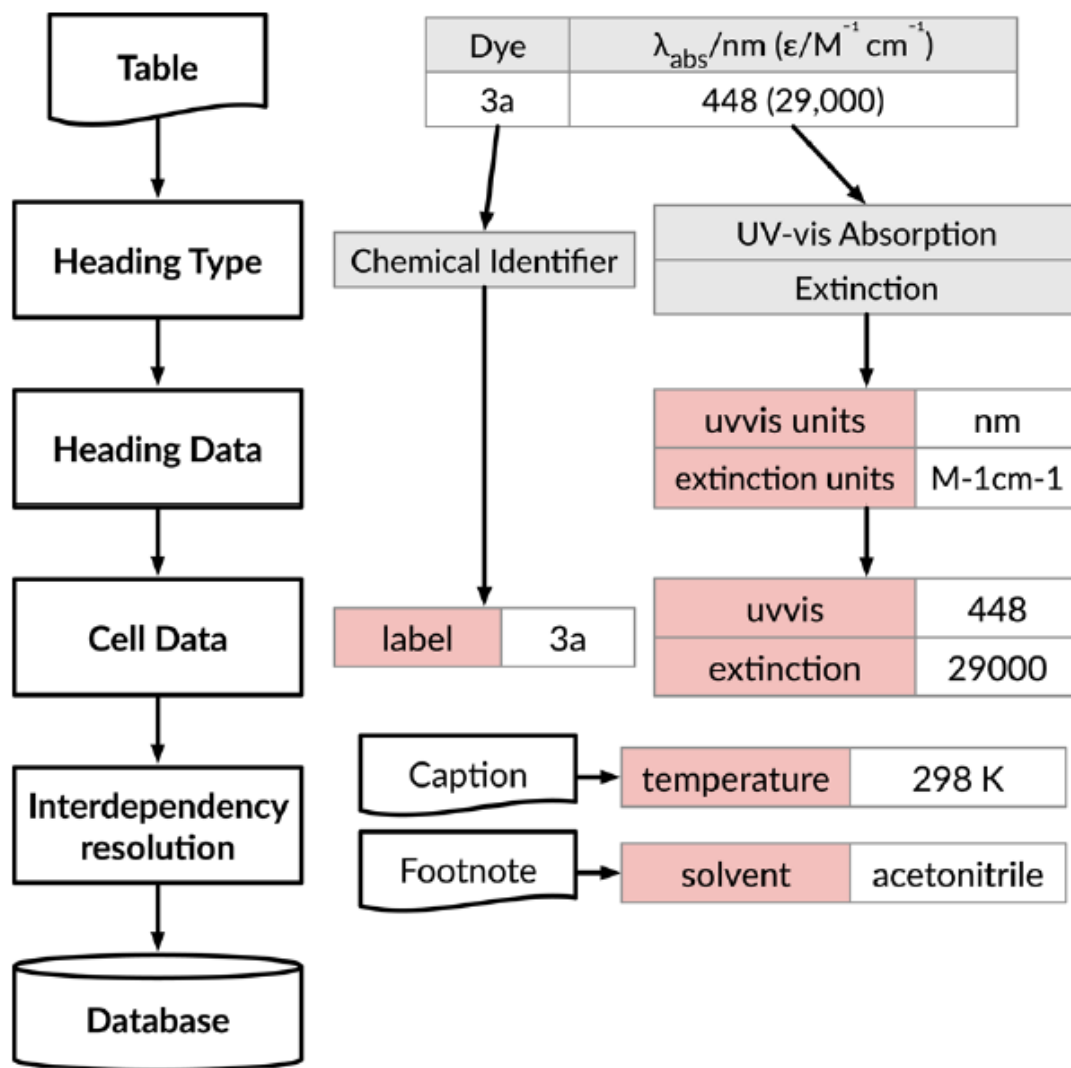
# ChemDataExtractor



# ChemDataExtractor



# ChemDataExtractor



# ChemDataExtractor

The dye 2-[2-[4-(dimethylamino)phenyl]diazenyl]-benzoic acid (**3a**) was added...

UV-vis spectra were recorded using an Agilent8453 diode array spectrophotometer.

```
{
  "name": "2-[2-[4-(dimethylamino)phenyl]diazenyl]-benzoic acid",
  "label": "3a",
  "uvvis": [ {
    "solvent": "acetonitrile",
    "apparatus": "Agilent8453 diode array spectrophotometer",
    "peaks": [ { "wavelength": "448", "extinction": "29,000" } ],
  } ],
}
```

**Figure 2:** UV-vis absorption spectra of 3a in acetonitrile.

Dye	$\lambda_{\max}/\text{nm}$ ( $\epsilon/\text{M}^{-1} \text{cm}^{-1}$ )
3a	448 (29,000)
3b	415 (48,000)

# ChemDataExtractor

$$\text{precision} = \frac{\text{TP}}{\text{TP} + \text{FP}}$$

$$\text{recall} = \frac{\text{TP}}{\text{TP} + \text{FN}}$$

$$\text{F-score} = 2 \cdot \frac{\text{precision} \cdot \text{recall}}{\text{precision} + \text{recall}}$$

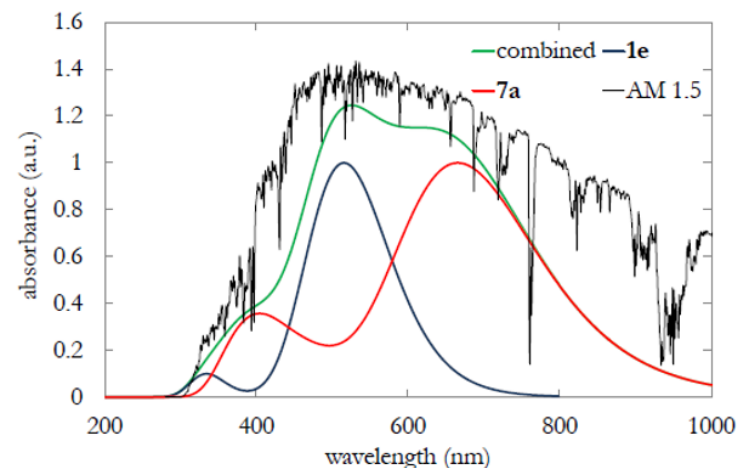
	precision	recall	F-score
chemical names	97.4%	96.3%	96.8%
alphanumeric labels	99.3%	97.3%	98.3%
<b>full chemical identifier records</b>	<b>94.1%</b>	<b>92.7%</b>	<b>93.4%</b>

	precision	recall	F-score
spectrum type	99.9%	96.7%	98.4%
chemical subject	93.4%	90.3%	91.8%
peak values	98.6%	95.4%	96.9%
solvent	99.5%	96.7%	98.1%
temperature	100%	87.5%	93.3%
apparatus	96.9%	91.0%	93.8%
<b>full spectrum records</b>	<b>88.3%</b>	<b>85.4%</b>	<b>86.8%</b>

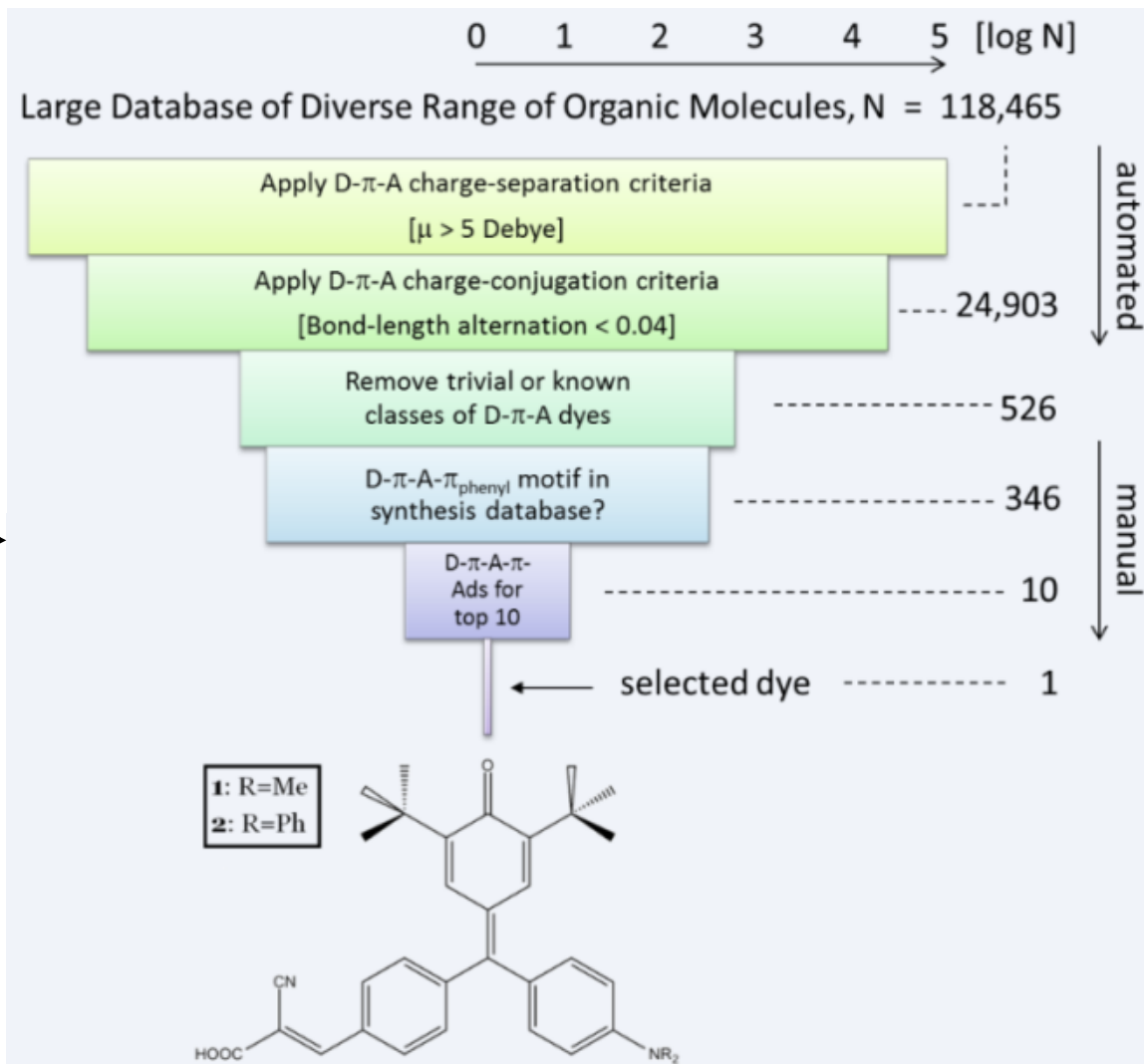
	precision	recall	F-score
property value	100%	95.9%	97.9%
property units	100%	94.8%	97.4%
chemical subject	93.5%	89.6%	91.5%
solvent	100%	94.4%	97.1%
temperature	100%	88.9%	94.1%
apparatus	100%	87.5%	93.3%
<b>full property records</b>	<b>93.5%</b>	<b>89.6%</b>	<b>91.5%</b>

	precision	recall	F-score
chemical identifier records	94.1%	92.7%	93.4%
spectrum records	88.3%	85.4%	86.8%
chemical property records	93.5%	89.6%	91.5%

# Materials discovery of new DSC dyes



Large database of UV/vis absorption spectra



# Materials Discovery 'Take #2'


## - The ADSP project

---

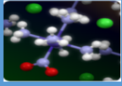
**Overview of Project:  
4 Key Steps**

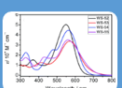
Academic Literature

**Apply/Create Software Tools for Data Extraction:**

 ChemDataExtractor

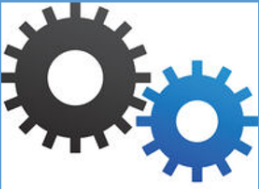
**Stage 1  
Data Extraction**

 **Material structure**

 **Physical properties**

**Stage 2  
Data Enrichment**

- Machine learning
- Quantum-chemical calculations ( $\varphi$ )




**Data Sources**



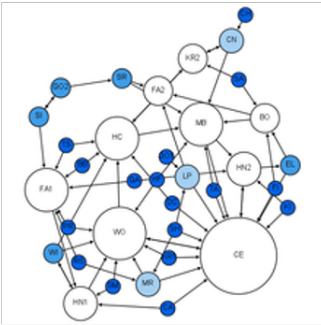
**Feedback**

**Databases**

- Materials Prediction
- Experimental Validation



**Stage 4  
Prediction & Validation**



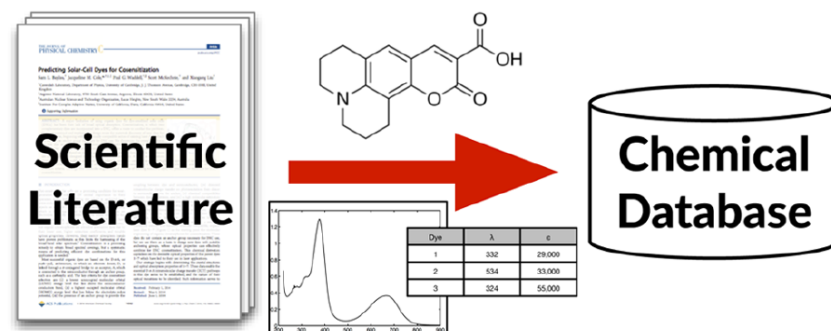
**Stage 3  
Create algorithms for materials prediction**



# Step 1: Data Extraction

- Negotiating publisher agreements

- Royal Society of Chemistry
- Elsevier
- Wiley
- Springer



- Parallelization of ChemDataExtractor code for Mira and Theta
- MongoDB for storing data

## Stage 2: Enriching Data with Machine Learning and Quantum-Chemical Calculations

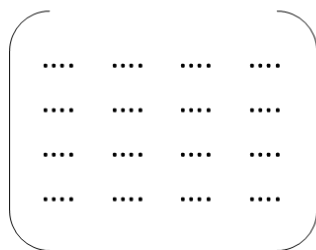
Chemical (SMILES)	property(1)	property(2)	Molecular descriptors
-----	----	----	10110110...
-----	----	----	10111110...
-----	----		11110101...
-----	----	----	11011010...

Missing Data

Training Data

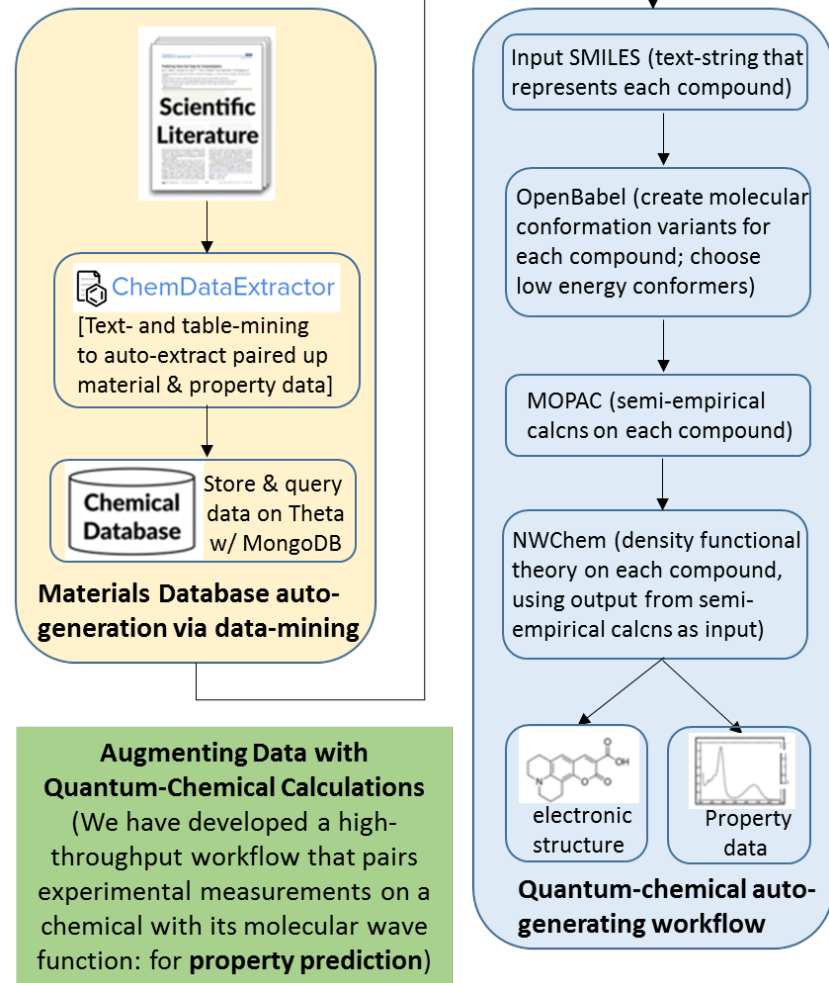
Machine learning:

Neural network,  
Random forest...



Feature vector

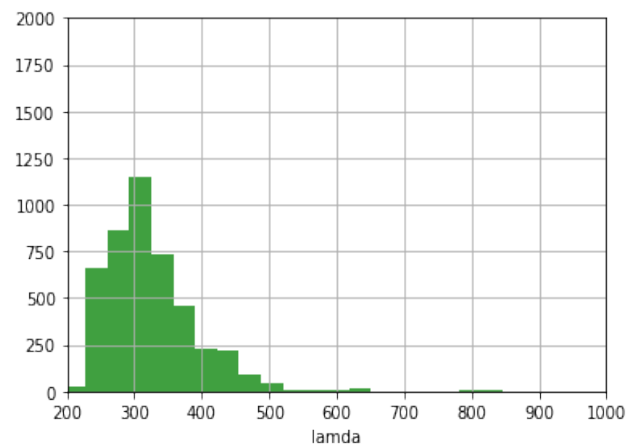
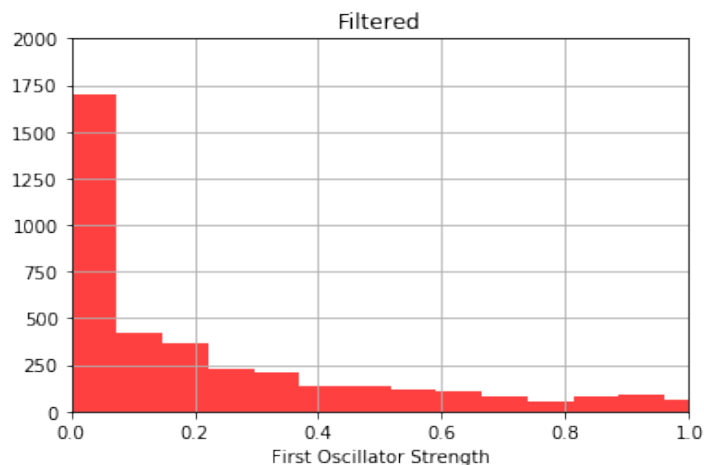
Predict Missing Data



# High-throughput Computation

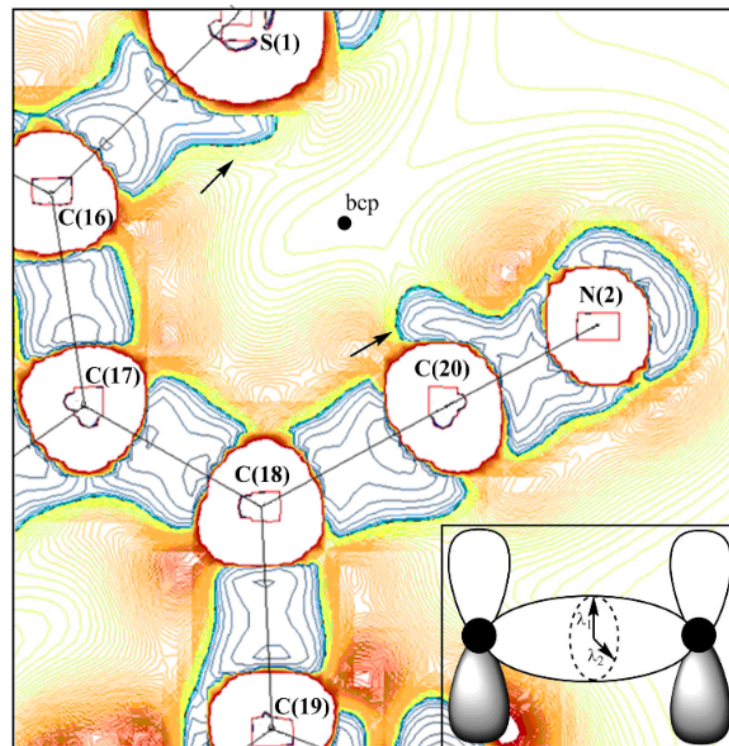
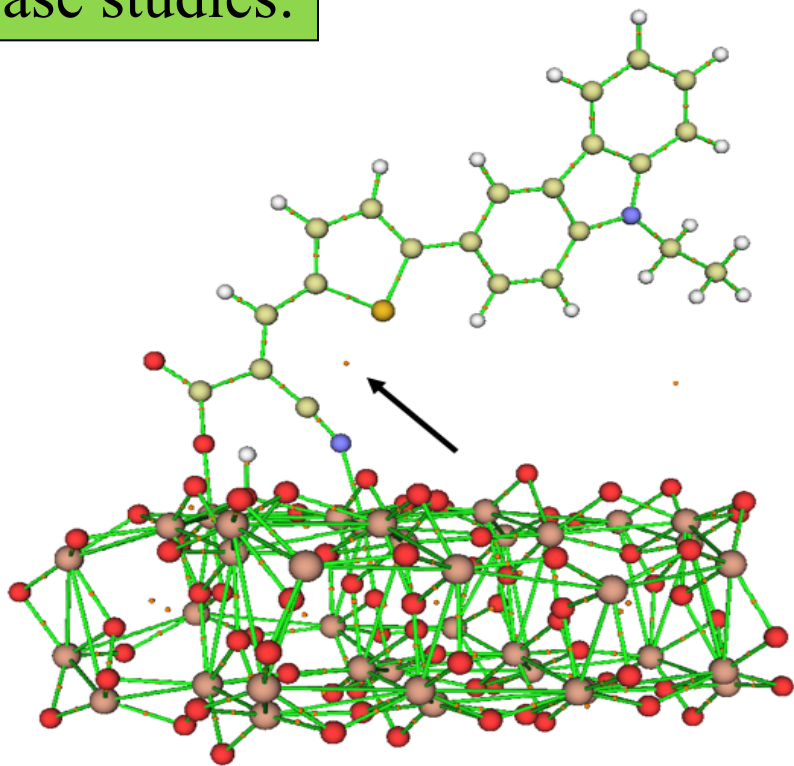
So far...

- 18,309 lambda max values from experiments
- 9,094 compounds.
- 4,519 computed with DFT and sTDA/sTDDFT .
- (sTDXX is an approximation to TDDFT)
- 917 possible candidates to be computed with Linear Response Coupled Cluster



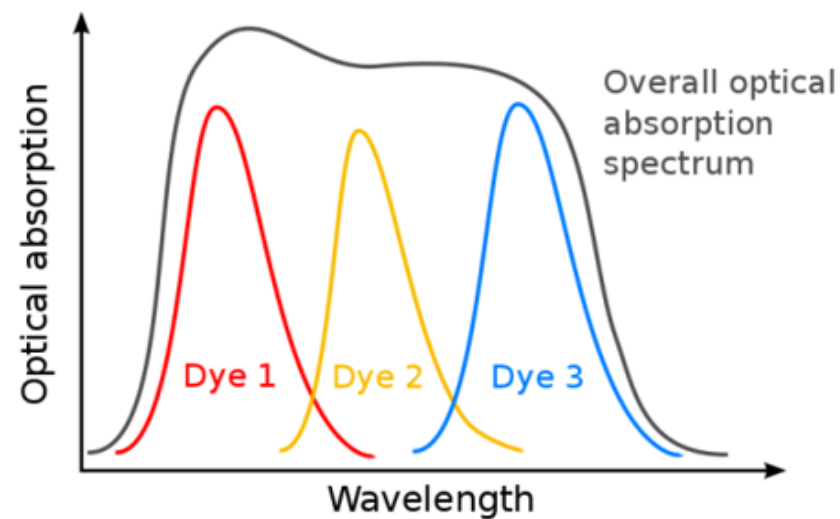
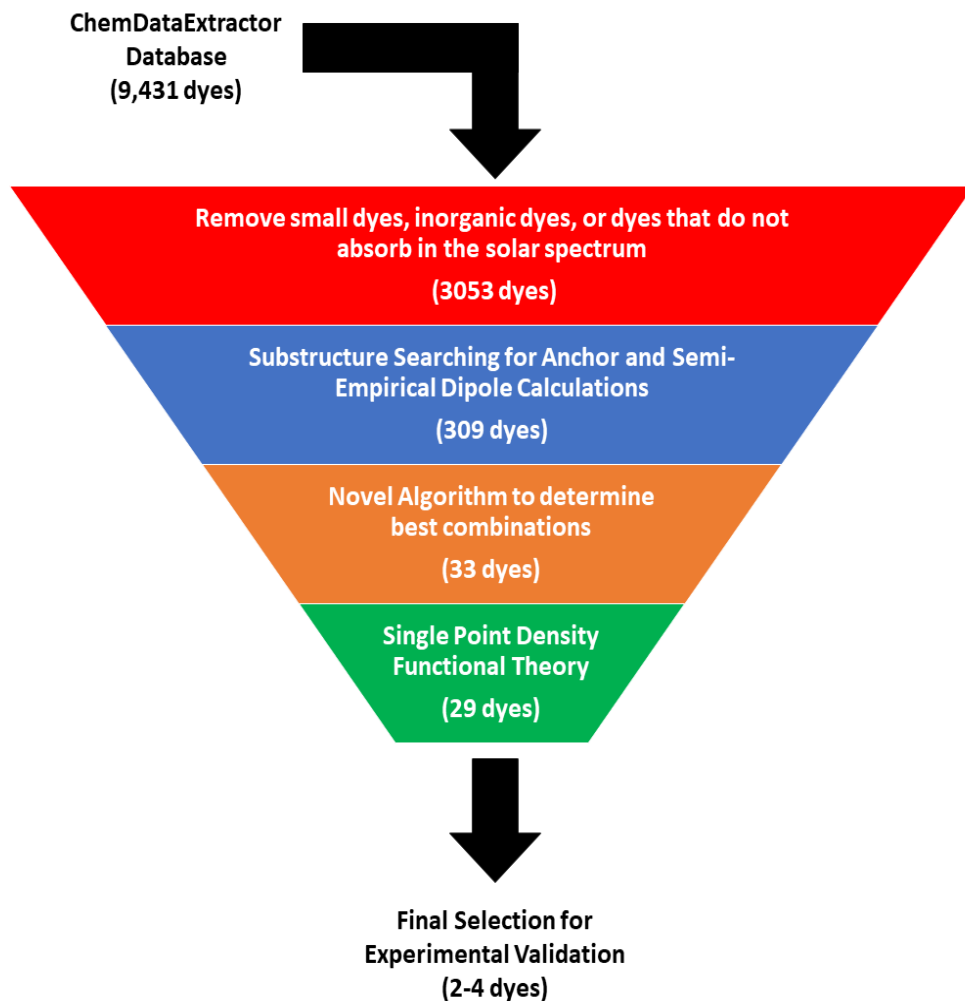
# Step 3: Design materials prediction

## Case studies:



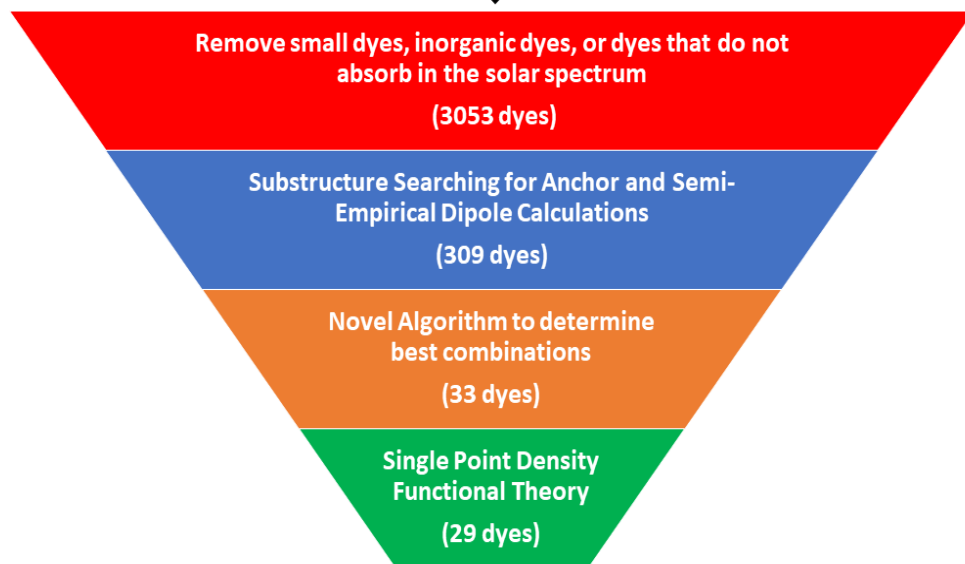
Cole, Vázquez-Mayagoitia et al, *ACS Appl. Mater. & Interfaces*, 9 (2017) 25952-25961.

# Step 3: Prediction workflows

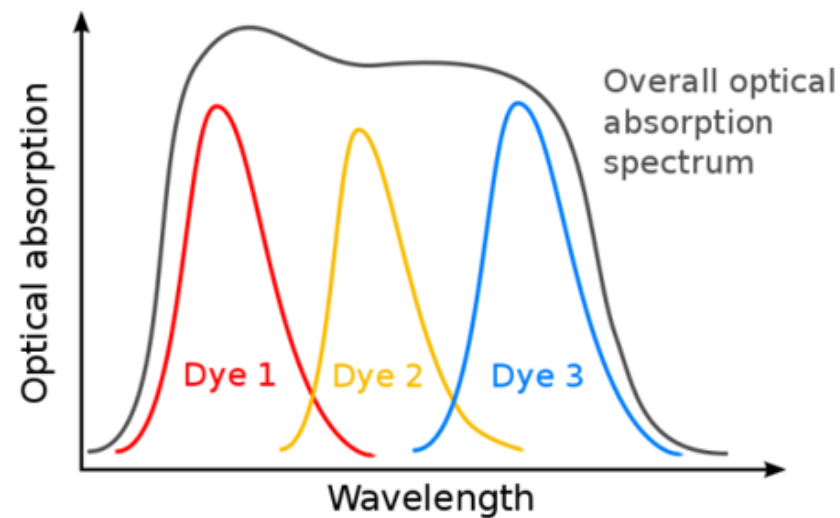


# Step 3: Prediction workflows

ChemDataExtractor  
Database  
(9,431 dyes)



Final Selection for  
Experimental Validation  
(2-4 dyes)



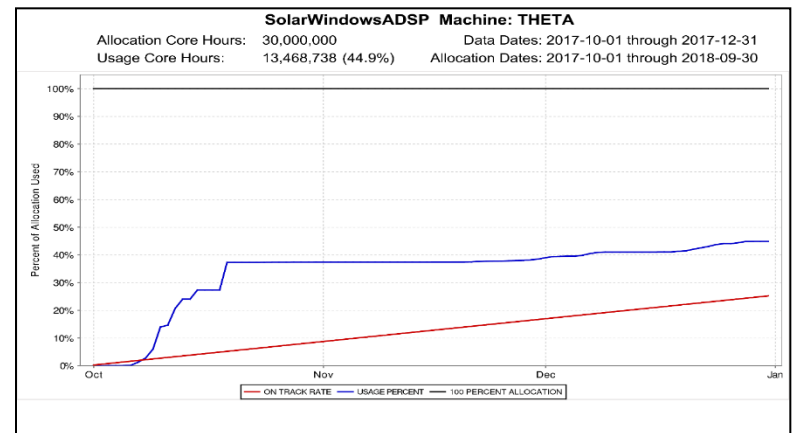
Step 4: Experimental Validation (NOW)

# Next Steps

- More Data: New Data Extraction capabilities:



- More Theta/Mira calculations
- Computation versus Experiment
- Data enrichment: Machine Learning



- Algorithm Design for Structure-Function relationships:
  - Case studies using DFT and MD
- Experimental Validation (CNM)

# Acknowledgements


Alvaro Vázquez-Mayagoitia, Ganesh Sivaraman, Samuel Flender, Yasi Ghadar, Elise Jennings, Venkat Vishwanath, Paul Coffman, Pragneshkumar Balvantbhai Patel, Misha Salim, User Support (ALCF) Petra Rudolf, Giota Stathi (Zernike Institute of Advanced Materials, Netherlands) Hiroaki Ozoe, K. Fuji, C.Kitamura, Takeshi Kawase (Uni. of Hyogo, Japan) Hiroyuki Kurata (University of Osaka, Japan) Chris Cooper, Kian Sing Low, Martin Blood-Forsythe, Yun Gong, Ed Beard, Matt Swain, Hao Chen (Cambridge University)



## References:

**Data-mining led DSC dye-discovery:**  
Cole et al, *PCCP*, 16 (2014) 26684

**Text- and table mining:**

Swain & Cole *J. Chem. Inf. Model.* 56 (2016) 1894-1904  [ChemDataExtractor](http://www.chemdataextractor.org)  
[www.chemdataextractor.org](http://www.chemdataextractor.org)